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國立臺灣大學99學年度碩士班招生考試試題

科目:有機化學(B)

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※注意:請於試卷上「非選擇題作答區」作答,並註明作答之題號。

- 1. (8%) Explain why (a) CO₂ has no dipole moment, but SO₂ does; (b) NH₃ has a much higher dipole moment than NF₃.
- 2. (8%) (a) Write the formula for the alkene hydrocarbon with the fewest number of carbons that exhibits both geometric and optical isomerism. (b) Provide the Fischer structural formulas for each stereoisomer.
- 3. (8%) Explain why (a) CCl₄ does not react with either Ag⁺ or OH⁻ and (b) SiCl₄ reacts vigorously with H₂O.
- 4. (12%) (a) Write the equations for preparing 1,3-butadiene from 1-butene. (b) Give the mechanism for the anionic (Nu:) induced polymerization of isoprene (2-methyl-1,3-butadiene). (c) As 1,3-penadiene and 1,4-pentadiene undergo hydrogenation, which one would produce higher heat of hydrogenation? Explain.
- 5. (8%) (a) Explain why benzene and naphthalene do not undergo a Diels-Alder reaction. (b) What is the structure of the Diels-Alder product of anthracene with ethene?
- 6. (6%) Write the equations for preparing 2,5-dimethylthiophene by heating a 1,4-dicarbonyl compound (provide one) with the appropriate inorganic reagent.
- 7. (12%) Bakelite is a phenol-formaldehyde resin formed from the base-catalyzed reaction of phenol and H₂C=O. (a) Give the product of the reaction of phenol with two equimolar amounts of H₂C=O. (b) Give the structure of the dimmers formed when the product in (a) is heated. (c) Further heating (curing) increases polymerization. Show the repeat unit in the final cured resin.
- 8. (6%) Write the equations for the preparation of HOOCCH₂CH₂COOH from CH₂=CH₂.
- 9. (20%) Draw the structure of the MAJOR product of the following reactions.

(a)
$$MeCH(CH_2COOH)_2 \xrightarrow{(CH_3CO)_2O}$$

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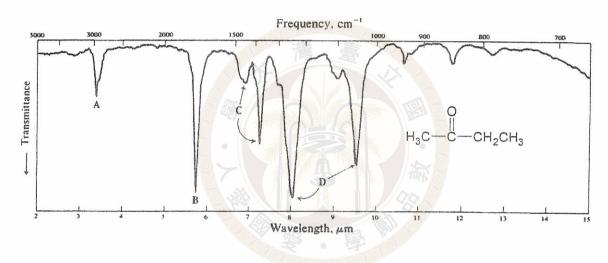
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(d)
$$Me_2N$$
 + $HCONMe_2$ $POCl_3$

10. (4%) Indentify the functional groups and the vibration modes for the peaks A-D in the following IR spectrum of ethyl acetate.



11. (8%) Draw the structure of the unknown compound with $C_{13}H_{16}O$ and ^{1}H NMR data (ppm): 7.54 (1H, d, J=16), 7.30 (2H, dd, J=2, 8), 7.21 (2H, dd, J=7, 8), 7.14 (1H, dd, J=2, 7), 6.67 (1H, d, J=16), 1.21 (9H, s).

試題隨卷繳回